

Critical Adsorption, Order-Parameter Profile, and Surface Tension: Theory Versus Experiment

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In the present work, we discuss the crossover behavior of the order parameter profile in simple fluids under critical and supercritical conditions. We develop a crossover theory for the critical adsorption profile, analogous to crossover theories for bulk critical phenomena in a simple fluid [1]. The crossover expressions are obtained to first order in ϵ within the RG approach proposed by Rudnick and Jasnow [2] for semi-infinite systems. The universal crossover function for the order-parameter profile is completely consistent with new Monte Carlo calculations on a three-dimensional Ising model [3] and in the asymptotic scaling regime corresponds to the Fisher-de Gennes hypothesis [4].

We test the theory against the surface excess (Gibbs) adsorption experimental data for $\text{CO}_2/\text{silicia}$ [5] and $\text{SF}_6/\text{graphite}$ [6] systems. On the non-critical isochores, good representation of experimental data is achieved in the range of temperatures from the saturated temperature up to $1.15T_c$ and densities $0.5\rho_c \leq \rho \leq 1.5\rho_c$. The optimization of the model to the excess isotherms in both systems indicates that they have surface critical behavior in the universality class of *normal* transitions. However, in this case the model does not reproduce the excess adsorption data along the critical isochore for $\text{SF}_6/\text{graphite}$ system at dimensionless temperatures $\tau \leq 0.01$. Analysis of the excess adsorption data along the critical isochore in $\text{SF}_6/\text{graphite}$ system indicates that the surface field h_1 vanishes linearly with τ as $T \rightarrow T_c$, which corresponds to the *ordinary* transition.

The crossover behavior of the surface tension in one-component fluids is also discussed. The crossover expression for the Sugden parameter was verified by direct comparison with the experimental data for Ar, O_2 , CH_4 , CO_2 , C_3H_{12} , C_6H_{14} , C_7H_{16} , and C_8H_{18} .

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